

0040-4039(95)02079-9

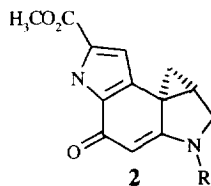
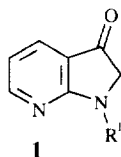
## Palladium Heteroannulation Process for Synthesis of Substituted Pyrrolo[2,3-*b*]Pyridin-3-ones

Eric Desarbre and Jean-Yves M  rour.\*

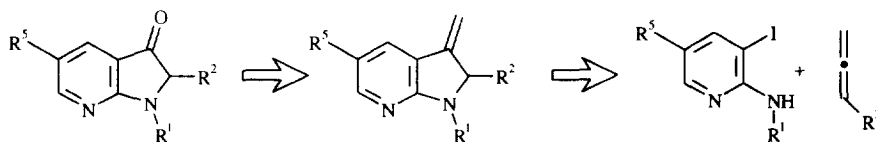
Institut de Chimie Organique et Analytique, associ   au CNRS, BP 6759,  
 Universit   d'Orl  ans 45067 Orl  ans Cedex 2, France.

**Abstract:** Substituted-3*H*-pyrrolo[2,3-*b*]pyridin-3-ones **7b-d** were prepared from 2-amino-3-iodopyridine derivatives **3b-d** by palladium carboannulation process with allenic compounds and then oxidative cleavage of the exocyclic carbon-carbon double bond.

Substituted-3*H*-pyrrolo[2,3-*b*]pyridin-3-ones or 7-azaindolinones **1** are versatile compounds for the preparation of azaindolic derivatives. As previously reported,<sup>1</sup> standard preparations of indolinones failed to yield 7-azaindolinones and we investigate new strategies to synthesise these compounds. We have already proposed a Baeyer-Villiger oxidation of 3-carboxaldehyde-1*H*-pyrrolo[2,3-*b*]pyridines<sup>1</sup> but the high cost of 7-azaindole and the limited possibilities of substitution (particularly in 5-position) lead us to investigate a new approach by creating the 7-azaindolic framework, by palladium heteroannulation, which can also be used in the synthesis of aza-analogues of Duocarmycins **2** (potent antitumor antibiotics).



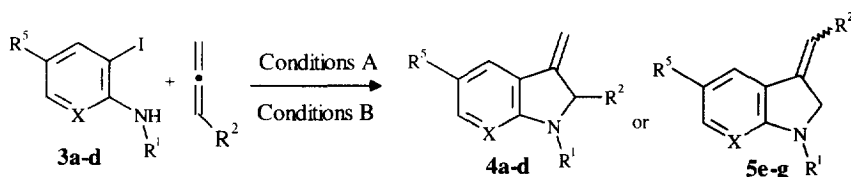
5-Substituted-3-iodo-2-aminopyridines are easily available by metalation reactions<sup>2</sup> and can be used for palladium-catalysed cross coupling reactions to provide 3-methylene-1,5-substituted-pyrrolo[2,3-*b*]pyridines as shown in the retrosynthetic scheme.



Larock has reported on the annulation process between iodo-aryl compounds and 1,2 dienes,<sup>3</sup> 1,3 dienes,<sup>4</sup> 1,4 dienes<sup>5</sup> and alkynes;<sup>6</sup> carbopalladation of allenes is well-documented;<sup>3,7</sup> the allenic compounds used were either alkyl substituted<sup>3</sup> or possessed an heteroatom.<sup>13b,c</sup> We report in this letter the

synthesis of 3-methylene-pyrrolo[2,3-*b*]pyridine derivatives using 1-substituted-1,2-propadienes and substituted 3-iodo-2-aminopyridines. The cyclisation, performed with *N*-tosyl-2-iodoaniline **3a** and 1-methoxypropadiene<sup>8</sup> in the presence of Pd(OCOCH<sub>3</sub>)<sub>2</sub> (conditions A) afforded exclusively the 1,2-dihydro-2-methoxy-3-methylene-1-tosyl indole **4a** with the same regioselectivity as reported for Larock's 1,2-undecadiene (scheme 1).<sup>3a</sup>

Switching for the azaanalogues, we obtained in the same way compounds **4b-d** using Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> and CH<sub>3</sub>CN as solvent (conditions B); the use of Pd(OCOCH<sub>3</sub>)<sub>2</sub> / PPh<sub>3</sub> and DMF as solvent (Conditions A) was unproductive (mainly degradation). Di *t*-butyl-1,2-propadienylphosphonate<sup>9</sup> (R<sup>2</sup> = PO[OC(CH<sub>3</sub>)<sub>3</sub>]<sub>2</sub>) gave the other regioisomers **5e,f** without substituent in 2-position (scheme 1).

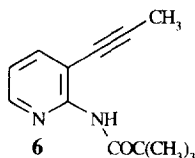


3	R <sup>1</sup>	R <sup>2</sup>	R <sup>5</sup>	X	Conditions	Reaction time	4,5	Yield of 4 (%)	Yield of 5 (%) ( <i>Z</i> / <i>E</i> )
<b>a</b>	SO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	OCH <sub>3</sub>	H	CH	A	5 h	<b>a</b>	75	-
<b>a</b>	SO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	OCH <sub>3</sub>	H	CH	B	5.5 h	<b>a</b>	60	-
<b>b</b>	SO <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	OCH <sub>3</sub>	H	N	B	3 h	<b>b</b>	65	-
<b>c</b>	COC(CH <sub>3</sub> ) <sub>3</sub>	OCH <sub>3</sub>	H	N	B	3 h	<b>c</b>	80	-
<b>d</b>	COC(CH <sub>3</sub> ) <sub>3</sub>	OCH <sub>3</sub>	Br	N	B	3 h	<b>d</b>	75	-
<b>a</b>	SO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>3</sub>	PO[OC(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	H	CH	B	14 h	<b>e</b>	-	50 (30 / 70) <sup>b</sup>
<b>c</b>	COC(CH <sub>3</sub> ) <sub>3</sub>	PO[OC(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	H	N	B	12 h	<b>f</b>	-	80 (30 / 70) <sup>a</sup>
<b>d</b>	COC(CH <sub>3</sub> ) <sub>3</sub>	PO[OC(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	Br	N	B	12 h	<b>g</b>	-	80 (30 / 70) <sup>b</sup>

Conditions A: 1 mmol of **3a**,<sup>11</sup> 2 mmol of allene, 0.05 mmol of Pd(OCOCH<sub>3</sub>)<sub>2</sub>, 0.05 mmol of PPh<sub>3</sub>, 1 mmol of C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>(CH<sub>3</sub>CH<sub>2</sub>)<sub>3</sub>N<sup>+</sup>Cl<sup>-</sup>, 3 mmol of Na<sub>2</sub>CO<sub>3</sub> in 4 ml of DMF are heated at 90°C during the time indicated. Conditions B: 1 mmol of **3a-d**,<sup>2</sup> 2 mmol of allene, 0.10 mmol of Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>, 1 mmol of C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>(CH<sub>3</sub>CH<sub>2</sub>)<sub>3</sub>N<sup>+</sup>Cl<sup>-</sup>, 3 mmol of Na<sub>2</sub>CO<sub>3</sub> in 4 ml of CH<sub>3</sub>CN are heated at 90°C during the time indicated. <sup>a</sup> isolated stereomers, <sup>b</sup> <sup>1</sup>H-NMR ratio.

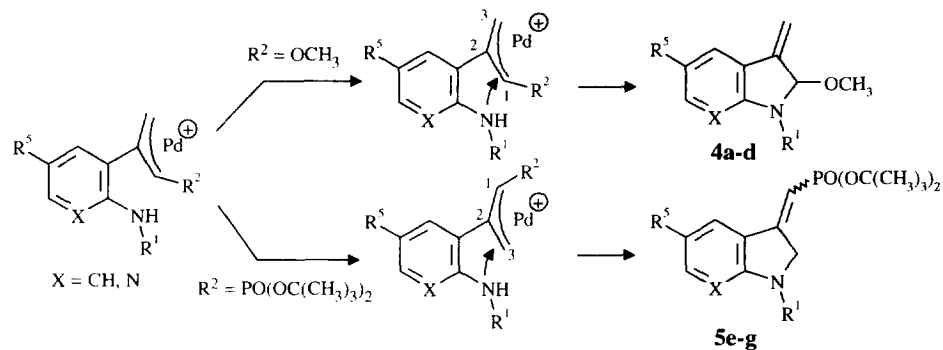
Scheme 1

Attempts with stannylallene (R<sup>2</sup> = Sn[(CH<sub>2</sub>)<sub>3</sub>CH<sub>3</sub>]<sub>3</sub>) and **3c** afforded 2,2-dimethyl-*N*-[3-(propyn-1-yl)pyridin-2-yl] propanamide **6** which results from the classic cross-coupling process involving stannyl compounds.<sup>10</sup> Degradation was observed with other allenic derivatives (R<sup>2</sup> = COOC<sub>2</sub>H<sub>5</sub>, C≡CCH<sub>2</sub>OH).

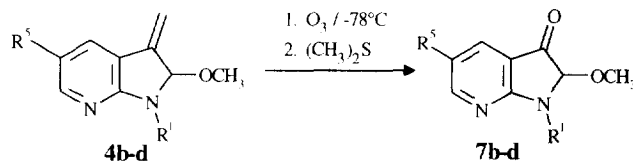


Two stereomers *Z/E* were obtained during the synthesis of **5e-g** and isomerisation of the minor stereomer *Z* to the *E*-isomer was observed in two days at room temperature.

Two different pathways can explain the outcome of the cyclisation. The first is the formation of a pyridinylpalladium compound followed by the  $\pi$ -allylic complexation of allenic derivative as reported for aryl derivatives.<sup>12</sup> Since the presence of polar substituents on terminal carbons of the  $\pi$ -allylic system influences the regiochemistry of the reactions,<sup>13a,b,c</sup> nucleophilic attack of nitrogen atom on the most electron-deficient carbon atom of the  $\pi$ -allyl system affords either **4** or **5**. With methoxyallene, the methoxy group stabilises the most electrophilic carbon, C-1 in this case.<sup>13b,c</sup> With allenic phosphonate, however the C-3 atom is the most positive, due to the electronic effect of the phosphonate group (scheme 2). Helquist has reported the same regioselectivity during palladium-catalyzed amination of 1,3-pentadienyl phosphonate ester.<sup>13d</sup>



Different methods were performed to cleave the carbon-carbon double bond of compounds **4a-d** into keto compounds **7a-d**; osmium tetroxide/sodium periodate,<sup>14</sup> heterogeneous permanganate oxidation<sup>15</sup> and ozonolysis<sup>16</sup>: the best method of oxidation was the ozonolysis reaction (scheme 3).



<b>4</b>	R <sup>1</sup>	R <sup>5</sup>	Reaction time	<b>7</b>	Yield of <b>7</b> (%)
<b>b</b>	SO <sub>2</sub> C <sub>6</sub> H <sub>5</sub>	H	0.15 h	<b>b</b>	25
<b>c</b>	COC(CH <sub>3</sub> ) <sub>3</sub>	H	0.3 h	<b>c</b>	54
<b>d</b>	COC(CH <sub>3</sub> ) <sub>3</sub>	Br	0.3 h	<b>d</b>	50

Ozonolysis: bubbling ozone, 0.40 mmol of **4b,c,d** in 3 ml of CH<sub>2</sub>Cl<sub>2</sub> at -78°C then addition of (CH<sub>3</sub>)<sub>2</sub>S.

Scheme 3

Some traces of 2-aminonicotinic are found during ozonolysis of compound **4c**; the oxidation of indolic compound **4a** with OsO<sub>4</sub>/NaIO<sub>4</sub> afforded a mixture of 1-tosyl-2-methoxy-2,3-dihydroindol-3-one (37% yield) and the corresponding 1,2-diol (37% yield).

The reactivity of the ketone function of the 7-azaindolinones **7b-d**, thus obtained, give access to 7-azaindole derivatives substituted in 3-position like 7-azatryptamines.<sup>1</sup> The bromine substituent in 5-position opens new avenue (metallation, Heck reaction) for the development of new active biological products. Furthermore, compounds **4c,d** are of great interest since the indolic analogues<sup>17</sup> unsubstituted in 2-position have been used in the synthesis of Duocarmycin A<sup>17c,e</sup> or Mitomycins.<sup>18</sup> The method here described give an easy and versatile access to 7-azaindole derivatives.

**Acknowledgement:** We thank J.C. Cintrat for NMR spectra data of compound **5e** and helpful discussions.

**Notes and references:**

- Desarbre, E.; Mérour, J.Y. *Tetrahedron Lett.* **1994**, *35*, 1995-1998.
- Estel, L.; Marsais, F.; Queguiner, G. *J. Org. Chem.* **1988**, 2740-2744.
- a) Larock, R.C.; Berrios-Peña, N.G.; Fried, C.A. *J. Org. Chem.* **1991**, *56*, 2615-2617. b) Larock, R.C.; Zenner, J. M. *J. Org. Chem.* **1995**, *60*, 482-483.
- a) Larock, R.C.; Berrios-Peña, N.G.; Narayanan, K. *J. Org. Chem.* **1990**, *55*, 3447-3450. b) Larock, R.C., Guo, L. *Synlett*, **1995**, 465-466.
- Larock, R.C.; Berrios-Peña, N.G.; Fried, C.A.; Yum, E.K.; Leong, W. *J. Org. Chem.* **1993**, *58*, 4509-4510.
- a) Larock, R.C.; Yum, E.K.; *J. Am. Chem. Soc.* **1991**, *113*, 6689-6690. b) Larock, R.C.; Yum, E.K.; Doty, J.M.; Sham, K.K.C. *J. Org. Chem.* **1995**, *60*, 3270-3271.
- a) Ahmar, M.; Cazes, B.; Goré, J. *Tetrahedron Lett.* **1984**, *25*, 4505-4508. b) Besson, L.; Bazin, J.; Goré, J.; Cazes, B. *Tetrahedron Lett.* **1994**, *35*, 2881-2884. c) Cazes, B. *Pure Appl. Chem.* **1990**, *62*, 1867-1878.
- Zimmer, R. *Synthesis* **1993**, 165-178.
- Glamkoski, E.J.; Gal, G. Purick, R.; Davidson, A.J.; Sletzing, M. *J. Org. Chem.* **1970**, 3510-3512.
- Mitchell, T.N. *Synthesis*, **1992**, 803-815.
- Luo, F.T.; Wang, R.T. *Heterocycles* **1991**, *32*, 2365-2372.
- a) Ahmar, M.; Barrioux, J. J.; Cazes, B.; Goré, J. *Tetrahedron* **1987**, *43*, 513-520. b) Kopola, N.; Friess, B.; Cazes, B.; Goré, J. *Tetrahedron Lett.* **1985**, *26*, 3795-3798.
- a) Collins, D.J.; Jackson, W.R.; Timms, R.N. *Tetrahedron Lett.* **1976**, 495-496. b) Vicart, N.; Cazes, B.; Goré, J. *Tetrahedron Lett.* **1995**, *36*, 535-538. c) Chaptal, N.; Colovray-Gotteland, V.; Grandjean, C.; Cazes, B.; Goré, J. *Tetrahedron Lett.* **1991**, *32*, 1795-1798. d) Nikaido, M.; Aslanian, R.; Scavo, F.; Helquist, P.; Åkermark, B.; Bäckvall, J.E. *J. Org. Chem.* **1984**, *49*, 4738-4740.
- Hayashi, T.; Kanehira, K.; Hagihara, T.; Kumada, M. *J. Org. Chem.* **1988**, *53*, 113-120.
- Lee, D.G.; Chen, T.; Wang, Z. *J. Org. Chem.* **1993**, *58*, 2918-2919.
- Griegee, R. *Angew. Chem., Int. Ed. Engl.* **1975**, *14*, 745-752.
- Exocyclic methylene compounds in 3-position of indoles were already prepared by other methods:
  - Ammonium ylide; Maeda, Y.; Shirai, N.; Sato, Y. *J. Chem. Soc. Perkin Trans. I* **1994**, 393-397. b) Chromium-Nickel complex; Hodgson, D. M.; Wells, C. *Tetrahedron Lett.* **1994**, *35*, 1601-1604. c) Zirconocene-stabilized complex; Tidwell, J.H.; Buchwald, S.L. *J. Am. Chem. Soc.* **1994**, *116*, 11797-11810. d) Palladium coupling with propynyl iodo aniline; Ref. 11. e) Palladium coupling with vinyl iodo aniline; Sakamoto, T.; Kondo, Y.; Uchiyama, M.; Yamanaka, H. *J. Chem. Soc. Perkin Trans I* **1993**, 1941-1942.
- Danishefsky, S.J.; Schkeryantz, J.M. *Synlett* **1995**, 475-490.
- Selected spectra data **4b** and **7b**: **4b**: m.p.: 72 - 74°C (CH<sub>3</sub>OH / H<sub>2</sub>O) ; I.R. (KBr)  $\nu = 1650, 920 \text{ cm}^{-1}$ ; <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300Mhz)  $\delta$  ppm: 1.50 (s, 9H, 3 x CH<sub>3</sub>), 3.31 (s, 3H, OCH<sub>3</sub>), 5.51 (d, 1H, H<sub>eth</sub>, J = 1.5 Hz), 5.78 (d, 1H, H<sub>eth</sub>, J = 1.5 Hz), 6.26 (m, 1H, CH), 6.92 (dd, 1H, H<sub>5</sub>, J = 5.2, 7.4 Hz), 7.67 (dd, 1H, H<sub>4</sub>, J = 1.5, 7.4 Hz), 8.20 (dd, 1H, H<sub>6</sub>, J = 1.5, 5.2 Hz). **7b**: m.p.: 58 - 60°C (ether); I.R. (KBr)  $\nu = 1740, 1680 \text{ cm}^{-1}$ ; <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300Mhz)  $\delta$  ppm: 1.51 (s, 9H, 3 x CH<sub>3</sub>), 3.62 (s, 3H, OCH<sub>3</sub>), 5.52 (s, 1H, CH), 7.09 (dd, 1H, H<sub>5</sub>, J = 5.2, 7.4 Hz), 7.96 (dd, 1H, H<sub>4</sub>, J = 1.5, 7.4 Hz), 8.59 (dd, 1H, H<sub>6</sub>, J = 1.5, 5.2 Hz).

(Received in France 29 September 1995; accepted 30 October 1995)